

STEREOABLATIVE ENANTIOCONVERGENT CATALYSIS

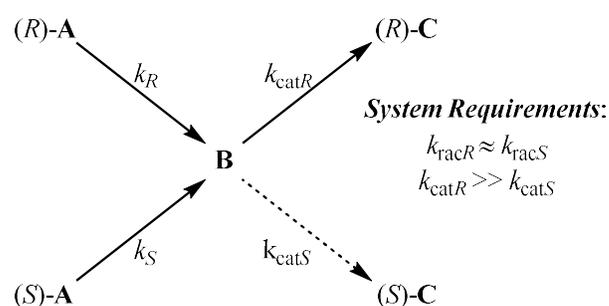
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INTRODUCTION

Synthetic chemists are continuously searching for new methods pertaining to the enantioselective construction of chiral molecules. To solve this problem, enantioconvergent catalysis, the complete conversion of a racemic substrate into a single, enantiomerically enriched compound, has arisen as an important strategy.¹ This area of catalysis has primarily been dominated by two approaches: dynamic kinetic resolution (DKR) and dynamic kinetic asymmetric transformation (DYKAT); while a third plausible tactic, stereoablative enantioconvergent catalysis, has considerably received less attention.

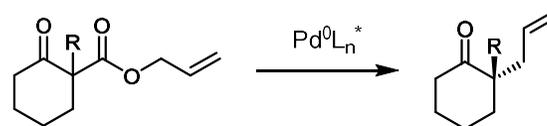
Operating under similar kinetic principles as DYKAT, stereoablative reactions do exactly as the name implies — a stereogenic center is irreversibly removed from the substrate, generating a prochiral compound, **B** (Figure 1). Notably, unlike DYKAT, there is no equilibration between the *R* and *S* enantiomers of



the racemic substrate and the prochiral compound. Upon interaction with the chiral catalyst, diastereomeric complexes are generated. Thereafter, a variety of processes can occur (e.g. nucleophilic or electrophilic attack, reductive elimination); however, in all cases, the rate of product formation will be faster for one of the diastereomeric complexes compared to the other.

ENOLATE-MEDIATED STEREOABLATION

Enolates are an ideal intermediate in stereoablative catalysis because generation of these species is well established. Arriving at this species in a creative manner,



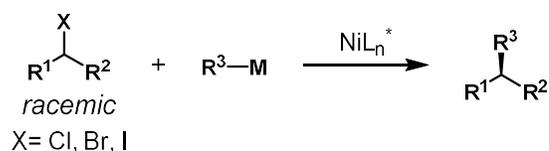
Stoltz and coworkers reported an enantioselective decarboxylative allylic alkylation in 2005 (Scheme 1).²

Scheme 1. Stoltz's enantioselective decarboxylative allylic alkylation (L_n^* = chiral ligand).

Using a β -keto-ester as the allyl source, this palladium-catalyzed transformation builds quaternary centers in high enantioselectivity. The reaction is versatile and has been extended to over 15 unique substrate classes.¹ In contrast to the Tsuji-Trost reaction, which often proceeds through external attack of the Pd- π -allyl species by a nucleophile, computational and experimental evidence has suggested that this transformation proceeds through an inner-sphere mechanism.³ Because of its significant synthetic utility, this transformation is often frequently employed in natural product synthesis.⁴

RADICAL-MEDIATED STEREOABLATION

A variety of enantioselective nickel-catalyzed, cross-coupling methods have been developed by Fu and coworkers that enable the direct, enantioselective synthesis of C_{sp3}-C_{sp3} bonds (Scheme 2). For these nickel-catalyzed couplings, mechanistic studies suggest that alkyl halide abstraction results in the formation of a



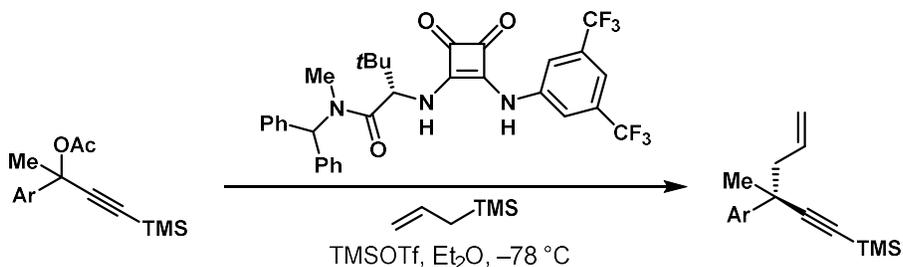
Scheme 2. Generalized system for Fu's enantioselective cross-couplings (L_n* = chiral ligand).

radical that is captured by a chiral catalyst.⁵⁻⁶ Most recently, the scope of these transformations has been expanded to include the cross-coupling of olefins with secondary and tertiary alkyl halides.⁷

CARBOCATION-MEDIATED STEREOABLATION:

When paired with an appropriate catalyst, carbocations can be exploited in stereoablative catalysis. In a key finding, Jacobsen and coworkers demonstrated that a chiral squaramide catalyst could build quaternary centers by an enantioconvergent allylation.⁸ Mechanistic studies suggest a stereoablative pathway by the formation of a

carbocation, suggesting that an S_N1-type reaction might be operative. Although this area is still in its infancy, these initial results are promising leads for future development.



Scheme 3: Jacobsen's S_N1-type allylation

References

1. Bhat, V.; Welin, E. R.; Guo, X.; Stoltz, B. M. *Chem. Rev.* **2017**, *117*, 4528.
2. Mohr, J. T.; Behenna, D. C.; Harned, A. M.; Stoltz, B. M. *Angew. Chem. Int. Ed.* **2005**, *44*, 6924.
3. Keith, J. A.; Behenna, D. C.; Sherden, N.; Mohr, J. T.; Ma, S.; Marinescu, S. C.; Nielsen, R. J.; Oxgaard, J.; Stoltz, B. M.; Goddard, W. A., III. *J. Am. Chem. Soc.* **2012**, *134*, 19050.
4. Enquist, J. A., Jr.; Stoltz, B. M. *Nature* **2008**, *453*, 1228.
5. Schley, N. D.; Fu, G. C. *J. Am. Chem. Soc.* **2014**, *136*, 16588.
6. Fischer, C.; Fu, G. C. *J. Am. Chem. Soc.* **2005**, *127*, 4594.
7. Wang, Z.; Yin, H.; Fu, G. C. *Nature* **2018**, *563*, 379.
8. Wendlandt, A. E.; Vangal, P.; Jacobsen, E. N. *Nature* **2018**, *556*, 447.