

## Supramolecular Pillars in Stimuli Responsive Vesicles

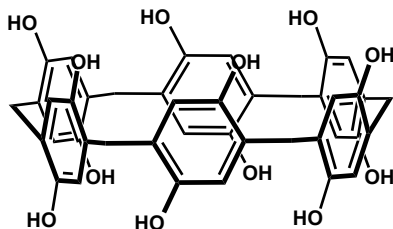
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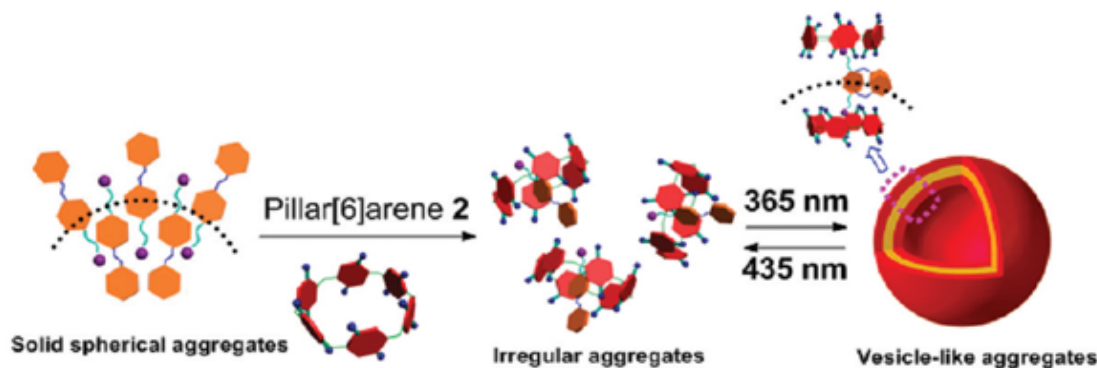
Controlled release technologies have received considerable attention throughout the years for their applicability in fields of such as adhesives, drug delivery, visual indicators and pesticide/herbicide distribution. Vesicles are one type of transport vehicle that is intrinsically stimuli responsive because of its spontaneous non-covalent assembly. To achieve stimuli responsive vesicles, preparing amphiphilic chemical agents that are driven to self-assembly and have properties such as cleavable bonds, redox active groups, or undergo architectural reorganization through conformational changes that induces a disruption in vesicle formation. Examples of these types of chemical agents are supramolecular amphiphiles. Supramolecular amphiphiles are agents that form host-guest complexes through non-covalent interactions that lead to an assembly that possess a hydrophobic and a hydrophilic domain. These systems are intrinsically stimuli responsive due to their non-covalent interactions bonding and therefore are gaining traction for this application.<sup>1</sup>

A new class of supramolecular macrocycles that are being applied to the construction of various assemblies such as, supramolecular polymers, molecular machines and artificial transmembrane channels are pillar[n]arenes (P[n]As).<sup>2</sup> Introduced in 2008 by Ogoshi and co-workers, P[n]As are cyclic oligomers of para-linked hydroquinones. They are synthesized in a one step Friedel-Crafts type reaction of para-substituted hydroquinone with paraformaldehyde in the presence of a Lewis base resulting in a columnar structure with a ridged hollow electron-rich interior (Figure 1).<sup>3</sup>



**Figure 1.** Three-dimensional representation of the chemical structure of Pillar[6]arene.

These hollow pillar-like structures yield macrocycles of 5 to 15 hydroquinone units in length. P[n]As and have been shown to complex with a variety of cationic organic salts such as trimethyl ammoniums, pyridiniums and paraquat.<sup>4</sup> Their ability to form strong host-guest complexes has prompted the use of P[n]As in the development of stimuli responsive vesicles. In 2012, Huang and co-workers were the first to report the synthesis of a pillararene system that would form vesicles when an ammonium azobenzene derivative formed the proper host-guest interactions. This system was demonstrated to undergo hierarchical reorganization when a photo stimulus was applied and prompted the development of a variety of stimuli responsive pillararene vesicle systems (Figure 2).<sup>5</sup>



**Figure 2.** Schematic representation of P[*n*]A photo-responsive vesicles.

Although Huang's work demonstrated that P[*n*]A systems could be used to form vesicles in a supramolecular fashion, one limitation to the system was the lack of water solubility. This led to the use of water-soluble ionically functionalized P[*n*]As as a strategy that was used by other groups to build amphiphilic P[*n*]A systems.<sup>6</sup>

Since Huang's work, P[*n*]A systems that are responsive to other stimuli such as pH, enzymatic cleavage, and redox conditions have been employed.<sup>7-10</sup> The combination of water soluble P[*n*]As combined with the use of biocompatible stimuli allows these vesicles to be used in drug delivery application. Using a redox responsive system, P[*n*]A vesicles were shown to encapsulate doxorubicin, penetrate in to cells and affect viability.

P[*n*]A host-guest systems have been shown to form vesicle-like structures with a stimuli responsive behavior. Their facile modification has allowed researchers to apply them to a variety of environments with a diversity of stimulus. Moving forward, the challenge for P[*n*]A supramolecular vesicles is to stand out above existing supramolecular systems one potential route for this would be the application of these systems to a set of unique chemical problems yet to be addressed by supramolecular vesicles.

## References

- (1) Loh, X. J. *Materials Horizons* **2014**, *1*, 185.
- (2) Song, N.; Yang, Y.-W. *Science China Chemistry* **2014**, *57*, 1185.
- (3) Ogoshi, T.; Kanai, S.; Fujinami, S.; Yamagishi, T.-a.; Nakamoto, Y. *Journal of the American Chemical Society* **2008**, *130*, 5022.
- (4) Tan, L.-L.; Yang, Y.-W. *J Incl Phenom Macrocycl Chem* **2014**, *1*.
- (5) Yu, G.; Han, C.; Zhang, Z.; Chen, J.; Yan, X.; Zheng, B.; Liu, S.; Huang, F. *Journal of the American Chemical Society* **2012**, *134*, 8711.
- (6) Xia, D.; Yu, G.; Li, J.; Huang, F. *Chemical communications* **2014**, *50*, 3606.
- (7) Zhou, J.; Chen, M.; Diao, G. *Chemical communications* **2014**, *50*, 11954.
- (8) Chang, Y.; Yang, K.; Wei, P.; Huang, S.; Pei, Y.; Zhao, W.; Pei, Z. *Angewandte Chemie* **2014**.
- (9) Li, Z.; Yang, J.; Yu, G.; He, J.; Abliz, Z.; Huang, F. *Organic letters* **2014**, *16*, 2066.
- (10) Yu, G.; Xue, M.; Zhang, Z.; Li, J.; Han, C.; Huang, F. *Journal of the American Chemical Society* **2012**, *134*, 13248.

