The Development and Application of the Asymmetric Pictet-Spengler Reaction

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# Introduction

In 1911, Amé Pictet and Theodor Spengler discovered Pictet and Spengler (1911): that  $\beta$ -phenylethylamine in presence of strong acid could condense with formaldehyde dimethyl acetal and cyclize to give tetrahydroisoquinoline(THIQ).<sup>1</sup> This reaction was expanded by Tatsui to include tryptamine derivatives to give tetrahydro- $\beta$ -carbolines(TH $\beta$ Cs) (Figure 1).<sup>2</sup> The TH $\beta$ C ring

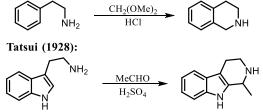


Figure 1. Discovery of Pictet-Spengler Reaction system is a privileged ring scaffold found in the indole alkaloid natural products which have important pharmaceutical properties.<sup>3</sup> The Pictet-Spengler reaction is the premiere way to synthesize these structures, so asymmetric variants are highly desirable.

## **Diastereoselective Pictet-Spengler Reaction:**

In 1979, Cook et. al. reported a diastereoselective Pictet-Spengler reaction between N-benzyl tryptophan methyl esters and bulky aldehydes. Cook also found that that mixtures of diastereomers underwent acid-catalyzed epimerization to give the *trans* product without eroding enantioenrichment.<sup>4</sup> In 2004, Bailey published the first synthetically useful *cis* selective Pictet-Spengler reaction between tryptophan allyl ester and aryl aldehydes.<sup>5</sup> A general *cis*-selective reaction with diverse aldehydes remains challenging. Similarly, stereogenic centers on the aldehyde partner can also direct the selectivity of the Pictet-Spengler reaction. Following condensation of an enantioenriched aldehyde with an amine substrate, the existing stereogenic center can control the addition of the indole group up into the imine.<sup>6a,6b</sup>

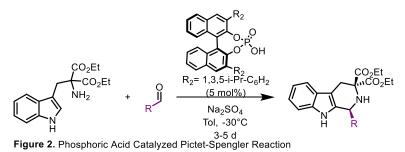
## **Chiral Auxiliary Strategies:**

To synthesize TH $\beta$ C's with a single stereogenic center on the ring, attaching chiral auxiliaries to the tryptamine substrates was done. Using N-phthaloyl amino acids to form the N-acyl iminium ion on a preformed imine, the resulting THBCs can be formed in good yield and diastereoselectivity.<sup>7</sup> This method has been applied to the total synthesis of quinolactacin A2 and B.<sup>8</sup>

### **Catalytic Enantioselective Pictet-Spengler Reactions**

In 2006, List et. al. reported a chiral Brønsted acid catalyzed Pictet- Spengler reaction. The method gave good yields and enantiomeric ratios but was hampered by a limited substrate scope (Figure 2).9In 2012, Wang showed that using SPINOL phosphoric acid catalysts could improve the yields and

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enantioselectivities for a variety of substrates. The system was still limited by the need for a bulky amine protecting group.<sup>10</sup>

In 2004, Jacobsen reported an N-acyl-Pictet-Spengler reaction catalyzed by

chiral thioureas using anion abstraction catalysis to make enantioenriched TH $\beta$ C. Forming the acyliminium ion increased the electrophilicity of the intermediate and allowed for the synthesis of TH $\beta$ Cs in 65-81% yield with up to 97.5:2.5 er.<sup>11</sup> Later Jacobsen et. al. demonstrated a non-acyl enantioselective

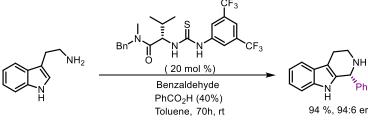


Figure 3. Jacobsen Enantioselective Catalytic Pictet- Spengler

Pictet-Spengler reaction using Brønsted Acid co-catalysis (Figure 3). Both aryl and aliphatic aldehydes are competent, and the TH $\beta$ Cs are accessed in good yields and enantiomeric ratios.<sup>12</sup> This method has been applied to

natural product synthesis for the total synthesis of Peganumine A.<sup>13</sup> In 2017, Jacobsen reported an mechanistic study.<sup>14</sup>

### **Conclusion:**

While there is still room for improvement, the asymmetric Pictet-Spengler reaction is a synthetically useful transformation for the synthesis of TH $\beta$ Cs. It has been used for total syntheses of indole alkaloids, and is used in industrial syntheses of pharmaceuticals.

### **References:**

- (1) Pictet, A.; Spengler, T. Ber. Dtsch. Chem. Ges. 1911, 44, 2030-2036
- (2) Tatsui, G. J. Pharm. Soc. Jpn. 1928, 48, 92
- (3) Zenk, M.; Juenger, M. Phytochemistry 2007, 68, 2757-2772
- (4) Cook et. al. J. Org. Chem. 1979, 44, 535
- (5) Bailey, P.D. et. al. Eur. J. Org. Chem. 2009, 1887-1890
- (6) (a) Stork, G. et. al. J. Am. Chem. Soc. 2005, 127, 16255-16262 (b) Zhao, G. et. al. Org. Lett. 2010, 12, 5366-5369
- (7) Waldmann, H.; Gunther Schmidt, D.; Henke, H.; Burkard, M. Angew. Chem. Int. Ed. 1995, 34, 2402-2403
- (8) Zhang, X.; Jiang, W.; Sui, Z. J. Org. Chem. 2003, 68, 4523-4526
- (9) Seayad, J.; Seayad, A.M.; List, B. J. Am. Chem. Soc. 2006, 128, 1086-1087
- (10) Huang, D.; Xu, F.; Lin, X.; Wang, Y. Chem. Eur. J. 2010, 18, 3148-3152
- (11) Taylor, M.; Jacobsen, E. J. Am. Chem. Soc. 2004, 126, 10558-10559
- (12) Klausen, R.; Jacobsen, E.Org, Lett. 2009, 11, 887-890
- (13) Piemontesi, C.; Wang, Q.; Zhu, J. J. Am. Chem. Soc. 2016, 138, 11148-11151
- (14) Jacobsen, E. et. al. J. Am. Chem. Soc. 2017, 139, 12299-12309