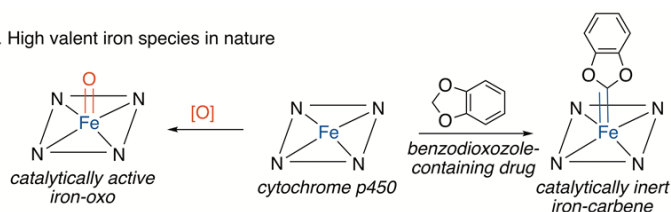


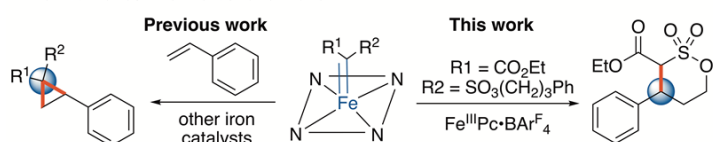
Catalytic C(sp³)-H Alkylation via an Iron Carbene Intermediate

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A. High valent iron species in nature



B. Iron carbenes in C–C bond formation



catalyzed alkylation of allylic and benzylic C(sp³)-H bonds. Mechanistic investigations support that an electrophilic iron carbene mediates homolytic C–H cleavage and rebounds from the resulting organoiron intermediate to form the C–C bond; both steps are tunable via catalyst modifications. These studies suggest that for iron carbenes, distinct from other late metal carbenes, C–H cleavage is partially rate-determining and must be promoted to effect reactivity.

The catalytic transformation of a C(sp³)-H bond to a C(sp³)-C bond via an iron carbene intermediate represents a long-standing challenge. Despite the success of enzymatic and small molecule iron catalysts mediating challenging C(sp³)-H oxidations and aminations via high-valent iron oxos and nitrenes, C(sp³)-H alkylations via isoelectronic iron carbene intermediates have thus far been unsuccessful. Iron carbenes have been shown to favor olefin cyclopropanation and heteroatom-hydrogen insertion. Herein we report an iron phthalocyanine-

Application of eNTRY Rules for Small Molecule Accumulation in Gram-negative Bacteria to Generate a Broad-spectrum Antibiotic

Sarah Perlmutter and Paul J. Hergenrother

The continuing rise of multi-drug resistant Gram-negative bacteria is a global health concern, and new drug classes are necessary to treat these pathogens. However, no new structural classes of antibiotics for Gram-negative bacteria have been developed since the late 1960s, largely due the impermeable nature of their outer membrane. Until recently there has been a limited understanding of what properties allow small molecule accumulation in *E. coli*. The eNTRY rules developed by our lab state that compounds are likely to accumulate if they contain a non-sterically congested ionizable Nitrogen (primary amine), have low Three-dimensionality (Globularity ≤ 0.25), and are relatively Rigid (rotatable bonds ≤ 5). These rules are now being used to inform discovery of new classes of antibiotics for Gram-negative bacteria.

