

MOF-based Photocatalytic Reduction of CO₂

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The increased emission of CO₂ from burning fossil fuels has been causing serious issues such as global warming, and great efforts have been made on solving these problems.^{1,2} One potential method is direct conversion solar energy to chemical fuels, which is also known as artificial photosynthesis. With this method, the transformation of carbon dioxide to fuel and chemical feed stocks can be achieved.³ The great challenge for this process lies in optimizing efficiency, activity and selectivity. A number of catalytic systems have been designed, however, the combination of all three properties in one system is still absent.⁴ For example semiconductor-based photocatalysts have been used to accomplish the photocatalytic reduction with high performance, yet still suffer from weak CO₂ adsorption and activation.

Metal-organic frameworks (MOFs) are ideal platforms for achieve the combination of three properties in one system.^{5,6} Highly selective molecular catalysts within the backbone give the system high selectivity.^{7,8} With the extensive surface area and tunable pore sizes, active sites are easily accessible for substrates which leads to outstanding activity.^{9,10} Optoelectronic properties such as bandgaps and intrinsic charge carrier mobility have been engineered to achieve higher efficiency.^{11,12} In addition, high degree of stability in solution and tolerance to a wide range of pH condition, give MOF-based photocatalyst great potential.

$\{\text{Cu}_3(\text{TCA})_2(\text{dpe})_3(\text{H}_2\text{O})_3\}_n$ (MOF-Cu), $\{\text{Co}_3(\text{TCA})_2(\text{dpe})_3(\text{H}_2\text{O})_6\}_n$ (MOF-Co), and $\{\text{Ni}_3(\text{TCA})_2(\text{dpe})_3(\text{H}_2\text{O})_6\}_n$ (MOF-Ni), are isomorphous MOFs with different transition-metal centers with same organic linkages 4,4',4''-nitrotrimesic acid ligands (TCA) and 1,2-di(4-pyridyl)ethylene (dpe).¹³ These MOFs only contain a single active metal center, which makes them perfect models to study the influence of the different transition-metal involved in the catalytic reaction. Three MOFs showed very different performance in the photocatalytic reduction, and MOF-Ni stood out for its high specificity over two isomorphous MOFs. Furthermore, more insights about the mechanism (Figure 1) of the photocatalytic reduction were discovered.

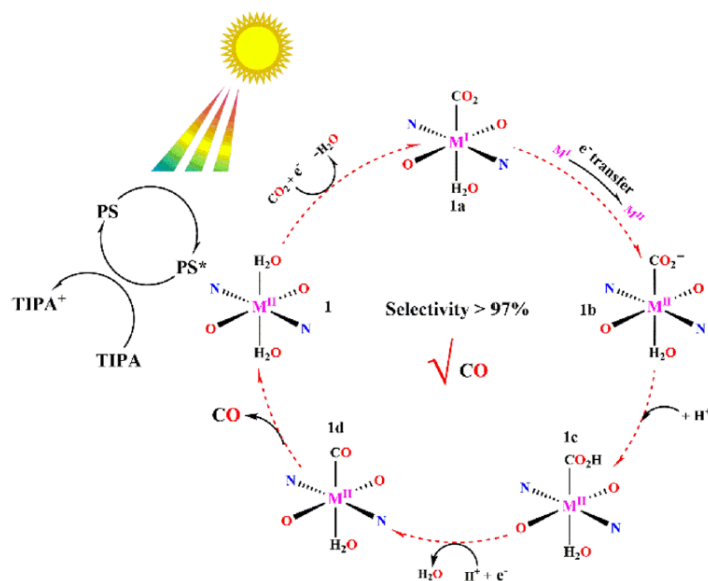


Figure 1. Proposed photocatalytic mechanism of MOFs for the CO₂ to CO conversion.¹³

BIF-20 (BIF = boron imidazolate framework), is constructed with porous boron imidazolate coordinated to tetrahedral Zn²⁺. Just like other other member from zeolite-like frameworks, BIF-20 has shown great ability for CO₂ storage and separation. With a rational combination of BIF-20 and a proper photosensitizer could not only tune the optical adsorption to desirable wavelength but also enhance the photocatalytic performance. Graphitic carbon nitride (g-C₃N₄) has been widely explored as photocatalyst for CO₂ reduction and it can be coupled with a porous structure to achieve higher performance. Thus g-C₃N₄ was integrated with BIF-20 and the new system has been proven to possess a much enhanced photocatalytic activity compare to g-C₃N₄ nanosheet (Figure 2).¹⁴

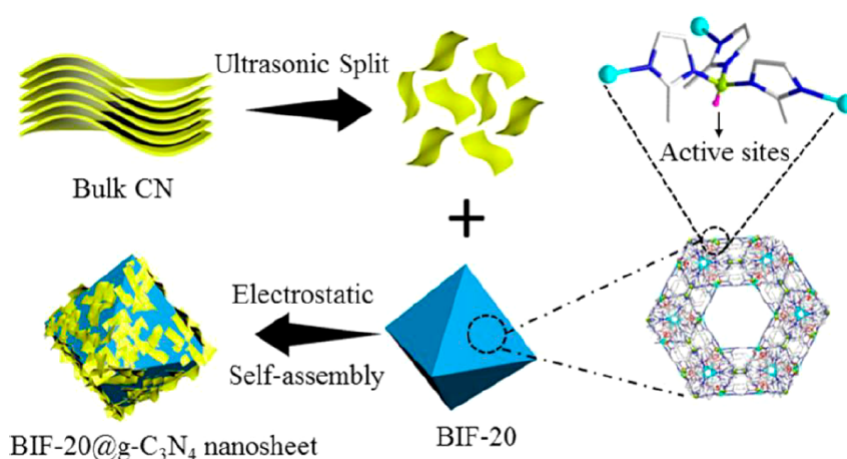


Figure 2. Schematic description of the preparation of the BIF-20@ g-C₃N₄ nanosheet photocatalyst¹⁴.

Halide perovskites (PVK) are optoelectronic materials featuring excellent optoelectronic properties such as large extinction coefficient and tunable absorption range, which allows them to be potential photocatalysts for CO₂ reduction. However, problems like low CO₂ capturing ability lead to low conversion efficiency. A PVK@MOF composite photocatalytic was designed and synthesis through a direct grow of zinc/cobalt-based zeolitic imidazolate framework (ZIF) on the surface of CsPbBr₃ quantum dots (Figure 3). This system has achieved better CO₂ capture ability and higher charge separation efficiency further lead to improved photocatalytic activity.¹⁵

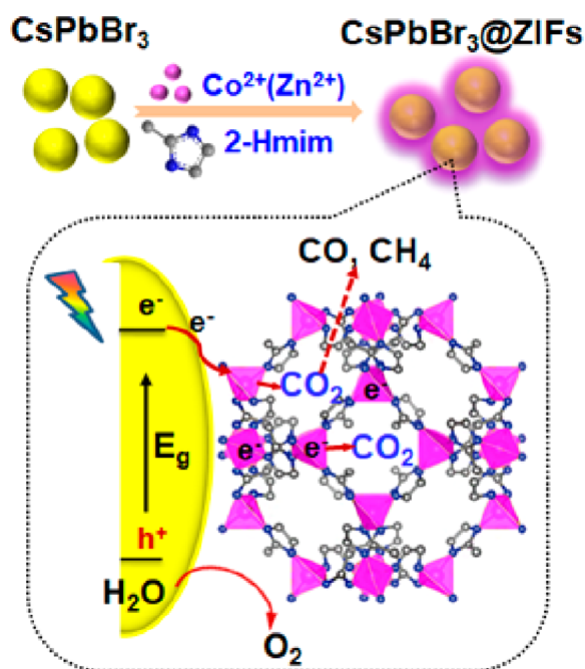


Figure 3. Schematic illustration of the synthesis process and CO₂ photoreduction process of CsPbBr₃/ZIFs¹⁵.

These examples show that MOFs can serve as a great platform to support known photocatalysts. Due to their structural and optoelectronic properties, the photocatalytic activity can be strongly enhanced compared to the catalyst alone. Quantum yield was barely mentioned in these papers, which might be the next step in developing MOF-based photocatalyst. Catalytic efficiency is one way to evaluate a catalyst, on the other hand, how effectively is energy input being used for the reduction is another concern.

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