Phosphinidene- and Arsinidene-Bridged Transition Metal Clusters

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Phosphinidenes [RP:] were first postulated to occur as reactive intermediates in the reactions of aryl- and alkyldichlorophosphines with magnesium or lithium in tetrahydrofuran, yielding cyclic or linear oligomers [1]. Schmidt and Osterroht in 1965 successfully trapped phenylphosphinidene, generated by heating tetraphenylcyclotetraphosphine above 160°C or reacting phenyldichlorophosphine with zinc dust at 25°C or oxidizing phenylphosphine with iodine, by insertion into the double bond of diethyldisulfide [2]. Similarly, arsinidenes [RAs:] were postulated by Nefedow and Manakow [3] to act as intermediates during cyclooligoarsane (RAs)n formation.

Huttner et. al [4] first reported the stabilization of an arsinidene, phenylarsinidene, in a metal complex in 1975. Soon thereafter, Huttner and coworkers reported similar phenylphosphinidene [5] and phenylstibinidene [6] complexes.

The unusual stability of these complexes arises from the nature of the arsinidene (or phosphinidene or stibinidene) ligand as a four-electron donor, as first described by Huttner [4]. The group VA atom completes its octet of electrons by metal  $d\pi$ -pnicogen  $p\pi$  back-bonding, while the ligand donates two electron pairs to the metal. The metal-group VA-metal bonding mode is most easily viewed as a  $4\pi$  electron-3 atom system.

 $\begin{array}{c} R \\ P \\ P \\ LMn \end{array} \xrightarrow{R} \\ MnL \end{array} \xrightarrow{R} \\ P \\ LMn \end{array} \xrightarrow{R} \\ P \\ MnL \end{array} \xrightarrow{R} \\ P \\ MnL \end{array} \xrightarrow{R} \\ P \\ MnL \end{array} \xrightarrow{R} \\ MnL \\ LMn \\ MnL \end{array}$ 

These complexes were the first examples of trigonal planar As(I) and P(I) coordination geometries. The "allylic" nature of these complexes is reflected in unusually short P-metal and As-metal bond distances, as determined by x-ray crystallography [5,7]. Fenske and Kostić recently reported [8] molecular orbital studies on  $[C_5H_5Mn(CO)_2]_2PC_6H_5$  which supported the three-center four electron bonding scheme in these complexes.

Inevitably, higher nuclearity phosphinidene, and to a lesser extent arsinidene, clusters have been prepared. Huttner and coworkers have prepared homometallic phosphinidene clusters from M<sub>3</sub>(CO)<sub>12</sub> where M = Fe [9], Ru [10], Os [11]. Geoffroy <u>et. al</u> [12] recently reported the preparation of a  $\mu_4$ -phosphinidene complex which is chemically "attractive" as it contains a catalytically active metal-Rh, labile 1,5-cyclooctadiene ligands, and is coordinately unsaturated. Other researchers have begun investigating mixed-metal phosphinidene clusters containing catalytically active metals such as Ru/Rh [13] and Ru/Au [13].

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Investigations into the catalytic behavior of phosphinidenebridged transition metal clusters have been scarce. Pittman et. al [14] has studied the hydroformylation of pentenes by  $Co_4(CO)_8$ - $(\mu_4-PC_6H_5)_2$  and Cook and Evans [15] have examined the tethering of  $(\mu-H)_2Ru_3(CO)_4\{\mu_3-PCH_2CH_2Si(OEt)_3\}$  on oxide supports. The catalytic potential of phosphinidene-bridged transition metal clusters is an area still open to extensive research.

Most recently, Huttner and coworkers have reported the preparation of new stibinidene complexes of manganese and chromium [16] as well as a phosphinidene chromium complex,  $(Cr(CO)_5)_2P(t-butyl)$ , with a <sup>31</sup>P-nmr chemical shift of 1362 ppm (rel. to H<sub>3</sub>PO<sub>4</sub>), reported by the authors as the largest <sup>31</sup>P-nmr chemical shift documented to date [17]. Marks et. al [18] have recently prepared and characterized the phosphinidene-bridged actinide methoxy complexes  $[Cp_2 M(OCH_3)]_2PH$ , where  $Cp' = n^5 - (CH_3)_5C_5$  and M = Th, U. Jones and Whittlesey [19] reported, this year, the formation of a chromium arsinidene complex which reacts further to form the heterometallacycle CrAs<sub>4</sub>. Finally Mathey and Marinetti [20] examined the cycloaddition of alkenes to a phosphinidene complex.

Phosphinidene- and arsinidene-bridged transition metal clusters containing a variety of metals have been prepared and structurally characterized. However, the reactivity, both stoichiometric and catalytic, of these complexes has only begun to be investigated and may exhibit a rich and varied chemistry.

## References

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